
Densities of active species in R/N₂ and R/N₂-x%H₂ (R = Ar or He) microwave early afterglows

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Résumé

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Thème 3 – Advances and challenges in plasma diagnostics

Flowing afterglows have previously been studied [1] in N₂ and Ar-N₂ gas mixtures to produce N₂ active species such as N-atoms, N₂(A), N₂(X, v > 13) metastable molecules and N₂⁺ ions. The interest of flowing afterglows is to select N₂ active species downstream the plasma, that is in conditions close to room temperature and without electric field. In long time afterglows (late afterglows), at times of 10⁻²-10⁻¹ s, the dominant N₂ active species are the N-atoms. At short times of 10⁻³-10⁻² s (early afterglows), N₂(A), N₂(X, v > 13) metastable molecules and N₂⁺ ions must be added to the N atoms.

Ar-N₂ gas mixtures were previously studied [1] as they are easier to ionize than pure N₂, producing a longer plasma column. It is presently compared the production of N₂ active species in the early afterglows of Ar-N₂ and He-N₂ microwave plasmas at a constant gas pressure of 8 Torr, a flow rate of 1slm in a dia. 18 mm post-discharge tube at distances of 3 and 20 cm (times 10⁻³-3 10⁻² s) and a transmitted power of 150 Watt. The interest of N₂ dilution into Ar and He is to increase the electron energy in the plasma by avoiding the electron vibrational barriers at 3 eV in pure N₂. The electron energy should be higher in He plasmas as a result of a lower electron excitation cross section (~ 3 10⁻¹⁸ cm²) compared with Ar (~ 3 10⁻¹⁷ cm²) and N₂ (~ 3 10⁻¹⁶ cm²) and a higher excitation threshold (19 eV in He and 12 eV in Ar).

A pre-mixed N₂-5%H₂ mixture was also introduced in place of pure N₂ to produce NH radicals and H-atoms in addition to the N₂ active species.

Absolute concentrations of active species are deduced from optical emission spectroscopy (OES) after NO titration and simplified chemical schemes. In R/N₂-x%H₂ (R = Ar or He)

*Intervenant

mixtures, the NH radical and H-atom densities are evaluated by considering that the excitation of the NH(A) radiative states in the afterglow is produced by $N_2(X, v > 13) + NH$ collisions.

It appears that the He/N₂-5%H₂ gas mixture could give about the same order of N and H-atom densities. The interest of these results concerns the enhancement of surface nitriding by combination of N and H atoms in afterglow conditions.

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